

## Original Article

# Evaluation of the Larvicidal and Repellency of Extracts and Silver Nanoparticles Biosynthesized with *Zataria multiflora* (Shirazi Thyme) against *Anopheles stephensi*

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## Abstract

**Introduction:** Mosquitoes transmit several diseases to humans. New and environmentally safe methods are needed to control mosquito populations effectively. This study aimed to evaluate the larvicidal and repellent properties of extracts and silver nanoparticles synthesized from *Zataria multiflora* (Shirazi thyme).

**Methods:** To determine the larvicidal and repellent properties of *Z. multiflora*, we tested different concentrations of synthesized silver nanoparticles and crude extract against this malaria vector.

**Results:** The crude extract exhibited weak larvicidal activity, with an LC<sub>50</sub> of 616.52 µg/mL and an LC<sub>90</sub> of 1109.74 µg/mL. However, the synthesized silver nanoparticles are highly active on the larvae of *Anopheles stephensi*, with an LC<sub>50</sub> at 51.07 µg/mL and an LC<sub>90</sub> at 119.04 µg/mL. Both the crude extract (ED<sub>50</sub>: 0.00031 µg/cm<sup>2</sup>; ED<sub>90</sub>: 0.0018 µg/cm<sup>2</sup>) and silver nanoparticles (ED<sub>50</sub>: 0.000072 µg/cm<sup>2</sup>; ED<sub>90</sub>: 0.00048 µg/cm<sup>2</sup>) demonstrated significant repellent activity. Particle size analysis revealed a range of 24.56 to 63.76 nanometers, with an average size of 52.7 nanometers.

**Conclusion:** This study showed that *Z. multiflora* silver nanoparticles have significant larvicidal properties, so they are suggested as a nature-friendly larvicide.

**Keywords:** Larvicidal; Repellency; *Zataria multiflora*; *Anopheles stephensi*; Silver nanoparticles

## Introduction

Mosquitoes transmit serious human parasitic and arboviral diseases such as malaria and dengue fever in tropical and subtropical regions (1). *Anopheles stephensi* transmits malaria in many regions of the world (2). *Plasmodium* parasites cause malaria. Among the different species, *P. falciparum* causes more deaths (2, 3). Vector control is a cornerstone preventing vector-borne diseases. The immature stages of mosquitoes are typically controlled using pyrethroids and organophosphates, while insecticide-treated nets (ITNs) and indoor residual spraying (IRS) are widely employed in tropical regions to curb malaria transmission (4, 5). While significant progress has been made in

controlling mosquito populations globally, the scarcity of effective insecticides remains a critical challenge. A promising approach to mosquito control involves combining insecticides with phytochemicals (6).

Nanoparticle biosynthesis offers an economical and environmentally friendly alternative for mosquito control, with advantages such as lower energy consumption, reduced chemical use, and minimal environmental impact (7). Currently, plants and microorganisms are used to synthesize nanoparticles (8). Silver nanoparticles (AgNPs) exhibit potent larvicidal and repellent activity against *Anopheles stephensi* (9). Advances in nanotechnology have enabled the

widespread production of silver nanoparticles (AgNPs) in laboratories globally (9). Nanotechnology advancements have led many laboratories worldwide to produce silver nanoparticles (AgNP). Various botanical combinations are currently used to control mosquitoes during outbreaks of mosquito-borne diseases. Therefore, plants possess excellent insect-repellent properties and low ecotoxicity (10–21). The repeated and widespread use of pesticides against mosquitoes has created and increased resistance in these health-threatened creatures. Therefore, it is necessary to search for effective and efficient control strategies (22–27). Nowadays, many studies have been conducted to investigate the insecticidal properties of plants. Several substances have been identified as effective insecticides, and some of them are currently under investigation for the creation of metal nanoparticles (28,29). Botanical nanoparticles exhibit mosquito-repellent and insecticidal properties even under challenging conditions (30–33).

Over 100 studies on metal nanoparticles have demonstrated their high effectiveness as repellents and mosquito-killing toxicants (2, 34, 35). *Zataria multiflora*, commonly known as Shirazi thyme or Iranian oregano, is an aromatic herb native to Iran and neighboring regions. It is widely used in traditional cuisine and herbal medicine due to its unique flavor and potential health benefits, including antimicrobial and antioxidant properties (35). In the current project, we surveyed the larvicidal and repellency characteristics of AgNP synthesized with *Z. multiflora* against the larvae and adults of *An. stephensi* respectively.

## Materials and Methods

### Plant sample

In the present study, fresh leaves of *Z. multiflora* (Shirazi Thyme) were collected from the mountains of Nurabad District with latitudes 30°6'25"N and longitudes 51°32'50"E, Fars Province, southern Iran, and subjected to

taxonomic identification in the Herbarium, Department of Pharmacognosy, and School of Pharmacy, Shiraz University of Medical Sciences (SUMS).

### Rearing mosquitoes

*Anopheles stephensi* larvae were reared in the Department of Biology and Vector Control of Diseases insectary at Shiraz University of Medical Sciences at a temperature of  $27 \pm 2$  °C, 80% humidity, and 10:14 photoperiod. Mosquito larvae were fed on fish flakes, and adult mosquitoes were fed on rabbit blood and 10% sucrose. Repellent experiments were conducted using 100 non-blood-fed female mosquitoes, aged 5–7 days.

### Extracting production, preparation of the nano solution

Shirazi thyme leaves were washed with water and dried at room temperature away from sunlight. To prepare the plant leaf broth, 12 grams of thyme leaves were mixed with 100 ml of water, boiled for 7 minutes, and filtered through filter paper. The filter was kept at -12 °C and used for testing within 7 days. The filter was treated with AgNO<sub>3</sub> (Aburaihan Pharmaceutical Company) aqueous solution (1–5 µm) and kept at room temperature. The final product was a yellow-brown, which showed the formation of Silver nanoparticles (AgNPs) with high stability in water.

### Characterization of Synthesized Silver Nanoparticles (AgNPs) Using SEM, EDS, FTIR, XRD, and UV-Vis Spectroscopy

Scanning electron microscopy (SEM), energy-dispersive spectroscopy (EDS), Fourier-transformed infrared (FTIR) spectroscopy, and X-ray diffraction (XRD) were used to analyze the attributes of synthesized AgNPs. A UV-Vis spectrophotometer was used to investigate the biological reduction of silver ions in solution. Scanning electron microscopy (SEM), mapping, Wavelength Dispersive Spectroscopy (WDS), and Electron Backscatter Diffraction (EBSD) were used to analyze the morphology and composition of the synthesized AgNPs.

tion (EBSD), equipped with Oxford Instruments©) FESEM /ZEISS /SigmaVP was used for nanoparticle size, morphology, and composition studies. X-ray diffraction (XRD) is primarily used to determine the crystal structure, phase identification, and crystallite size of materials. It is widely used in mineralogy, metallurgy, geology, and materials science to characterize crystalline materials such as metals, ceramics, minerals, and pharmaceuticals. Fourier-transformed infrared (FTIR) spectroscopy is used for qualitative and quantitative analysis of organic and inorganic materials. It is widely applied in chemistry, biochemistry, polymer science, pharmaceuticals, and forensic science for identifying compounds, characterizing molecular structures, and studying chemical reactions. In the analysis of synthesized AgNPs, EDS can confirm the presence of silver (Ag) in the nanoparticles and provide information about the elemental composition of any impurities or stabilizing agents present in the sample (3, 36).

### Testing the larvicidal effect

The larvicidal effect of *Z. multiflora* extract and its silver nanoparticles (AgNPs) was evaluated according to the WHO protocol (34). *Zataria multiflora* extracts were tested at concentrations of 240, 480, 720, 960, and 1200 µg/mL, and AgNPs synthesized using *Z. multiflora* extracts were tested at 24, 48, 72, 96, and 120 µg/mL concentrations. 249 ml of water and 1 ml of *Z. multiflora* leaf extract were mixed in a half-liter beaker, and then 24, 48, 72, 96, and 120 µg/mL of silver nanoparticles were added to this mixture. In the last step, four replications were performed for each concentration along with a control test. 20 larvae were used in each replication. Mortality of larvae was recorded after 24 hours of contact with different concentrations of crude extract and nanoparticles (38). The lethal concentration was calculated using probit regression analysis (39).

### Testing the repellency effect

To determine the effective repellency dose (ED), the Klun and Debboun model was used (40). In this method, a standard plastic cage consisting of a Plexiglas cube measuring 4x5x18 cm with four rectangular holes measuring 3 x 4 cm was used. To calculate the effective dose of each repellent, it was applied directly to volunteer human skin. The consent form was filled out for each volunteer. For the control test, N, N-diethyl-meta-toluamide (DEET) repellent was used as a positive control. Five circles were drawn on the skin, and 50 microliters of the diluted repellent were placed on them. The treated area was exposed to air to dry, and then an experimental apparatus containing five starved 3–5-day-old female mosquitoes per cell was attached to the treated skin, and the cell door was opened (41). The number of mosquito bites and landings was recorded every 5 minutes. To perform the test, each dose was repeated four times to estimate ED<sub>50</sub> and ED<sub>90</sub> (42). The concentrations used in *Z. multiflora* extract were 0.125, 0.25, 0.5, and 1 mg/cm<sup>2</sup>, and 0.125, 0.25, 0.5, and 1 mg/cm<sup>2</sup> for silver nanoparticles (AgNPs) synthesized using *Z. multiflora* extract. Protection time was evaluated for both compounds. All repellency tests were carried out in a standard laboratory setting. Mortality and repellency data were analyzed using probit analysis to calculate LC<sub>50</sub>, LC<sub>90</sub>, ED<sub>50</sub>, and ED<sub>90</sub> with 95% confidence limits.

## Results

During this study, a color change of *Z. multiflora* extract was observed when incubated with silver nitrate. After 70 minutes of incubation, the color of the extract changed from light brown to dark brown. The EDS plot was used for elemental identification, which showed that 18% of the compound was silver, confirming synthesis. Additionally, the EDS plot showed that the compound contains C (35.4%), O (21.5%), Mo (9%), K (5.3%), Cl (5.2%), Ca

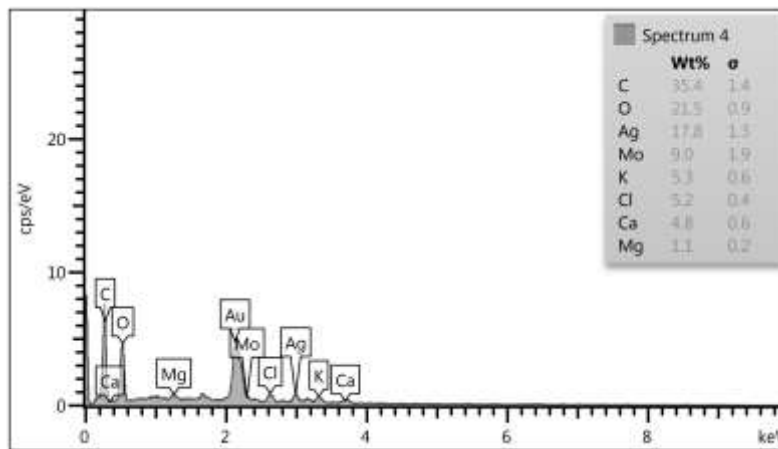
(4.8%), and Mg (1.1%) (Fig. 1). X-ray showed different peaks of silver nanoparticles synthesized from aqueous extracts of *Z. multiflora* leaves (Fig. 2). SEM revealed that the shape of AgNPs was spherical. This study showed that some particles were medium in size (52.7 nm) and some particles were small (24.56 nm), (zeta potential, -25.2 mV) (Fig. 3).

In this study, LC<sub>50</sub> and LC<sub>90</sub> for *Z. multiflora* extract were recorded as 616.52 µg/mL and 1109.74 µg/mL, respectively, while for synthesized silver nanoparticles, LC<sub>50</sub> and LC<sub>90</sub> were 51.07 and 119.04. µg/mL, respectively

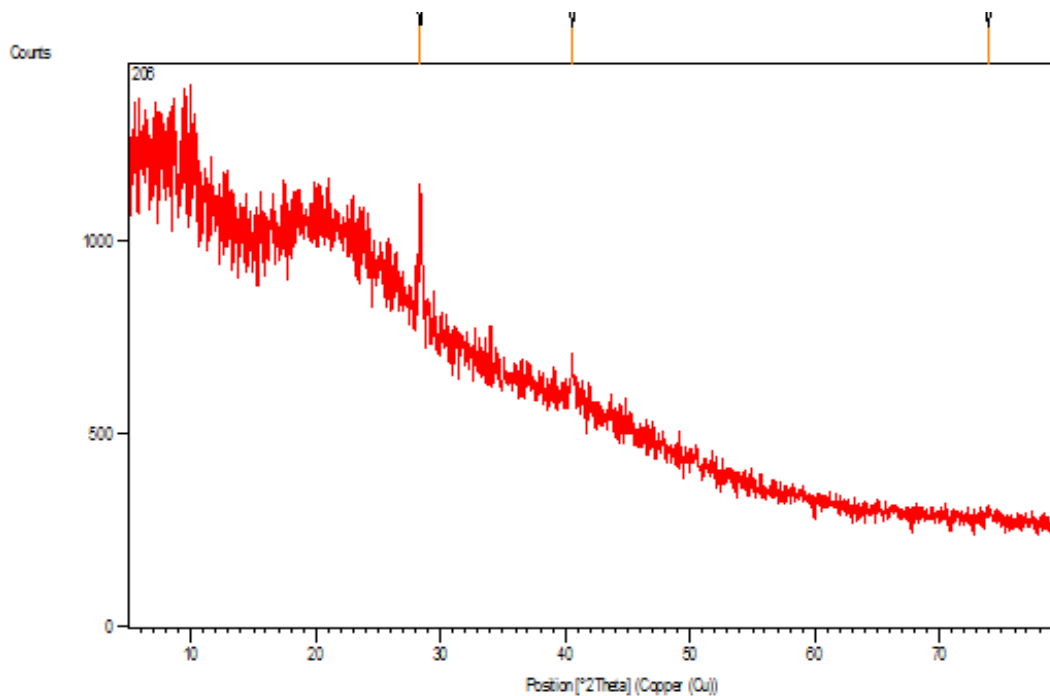
(Fig. 4 and Table 1). Using the K and D method, the results indicated that *Z. multiflora* extract exhibited significant repellent activity (ED<sub>50</sub>: 0.00031 µg/cm<sup>2</sup>; ED<sub>90</sub>: 0.0018 µg/cm<sup>2</sup>). Additionally, the repellent activity of silver nanoparticles (AgNPs) synthesized using *Z. multiflora* extract was recorded as ED<sub>50</sub>: 0.000072 µg/cm<sup>2</sup> and ED<sub>90</sub>: 0.00048 µg/cm<sup>2</sup>. Therefore, the results suggest that silver nanoparticles (AgNPs) are more effective in repelling compared to *Z. multiflora* extract. The protection time for the silver nanoparticle formulation varied between 20 and 115 minutes (Table 2).

**Table 1.** Larvicide rate of aqueous crude and synthesized AgNPs of *Zataria multiflora* extract against larvae of *Anopheles stephensi*, Shiraz City, Fars Province, Iran

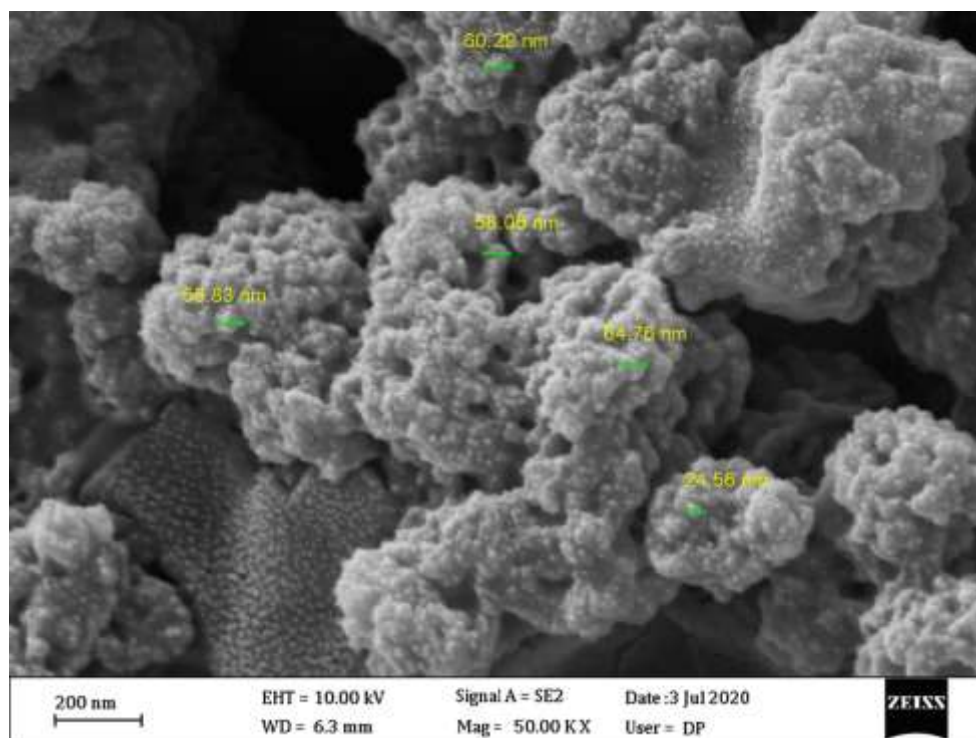
| Compound                        | Concentrations (µg/mL) | Mortality percentage %±SD | LC <sub>50</sub> (mg/L) (µg/mL) ±SD | LC <sub>90</sub> (mg/L) ±95 % CL | Heterogeneity x <sup>2</sup> | Df |
|---------------------------------|------------------------|---------------------------|-------------------------------------|----------------------------------|------------------------------|----|
| Crude aqueous extract           | 240                    | 17.5±2.1%                 | 616.52±0.75                         | 1109.74±45.23                    | 1.31                         | 2  |
|                                 | 480                    | 27.5±2.8%                 |                                     |                                  |                              |    |
|                                 | 720                    | 45.0±3.2%                 |                                     |                                  |                              |    |
|                                 | 960                    | 86.25±2.5%                |                                     |                                  |                              |    |
|                                 | 1200                   | 100±0.0%                  |                                     |                                  |                              |    |
| AgNPs <i>Zataria multiflora</i> | 24                     | 13.75±1.9%                | 51.07±0.26                          | 119.04±12.75                     | 2.20                         | 2  |
|                                 | 48                     | 46.25±3.5%                |                                     |                                  |                              |    |
|                                 | 72                     | 63.75±2.7%                |                                     |                                  |                              |    |
|                                 | 96                     | 87.5±2.2%                 |                                     |                                  |                              |    |
|                                 | 120                    | 100±0.0%                  |                                     |                                  |                              |    |



**Fig. 1.** Energy dispersive spectroscopy (EDS) analysis showing elemental composition of synthesized silver nanoparticles using *Zataria multiflora*. 5 µl of the sample was used for analysis. The results show that 18% of the compounds are silver, confirming the synthesis. Shiraz city, Fars province, Iran. X-axis: Energy (keV). Y-axis: Intensity (counts). Device: FESEM / ZEISS / Sigma VP equipped with Oxford Instruments EDS



**Fig. 2.** X-ray diffraction (XRD) pattern of silver nanoparticles synthesized from *Zataria multiflora* extract. Red line represents the fitted diffraction peaks indicating the crystalline structure of silver nanoparticles. No baseline control sample was used for comparison in this pattern. X-axis:  $2\theta$  degrees. Y-axis: Intensity (arbitrary units). Device: PANalytical X'Pert PRO XRD system

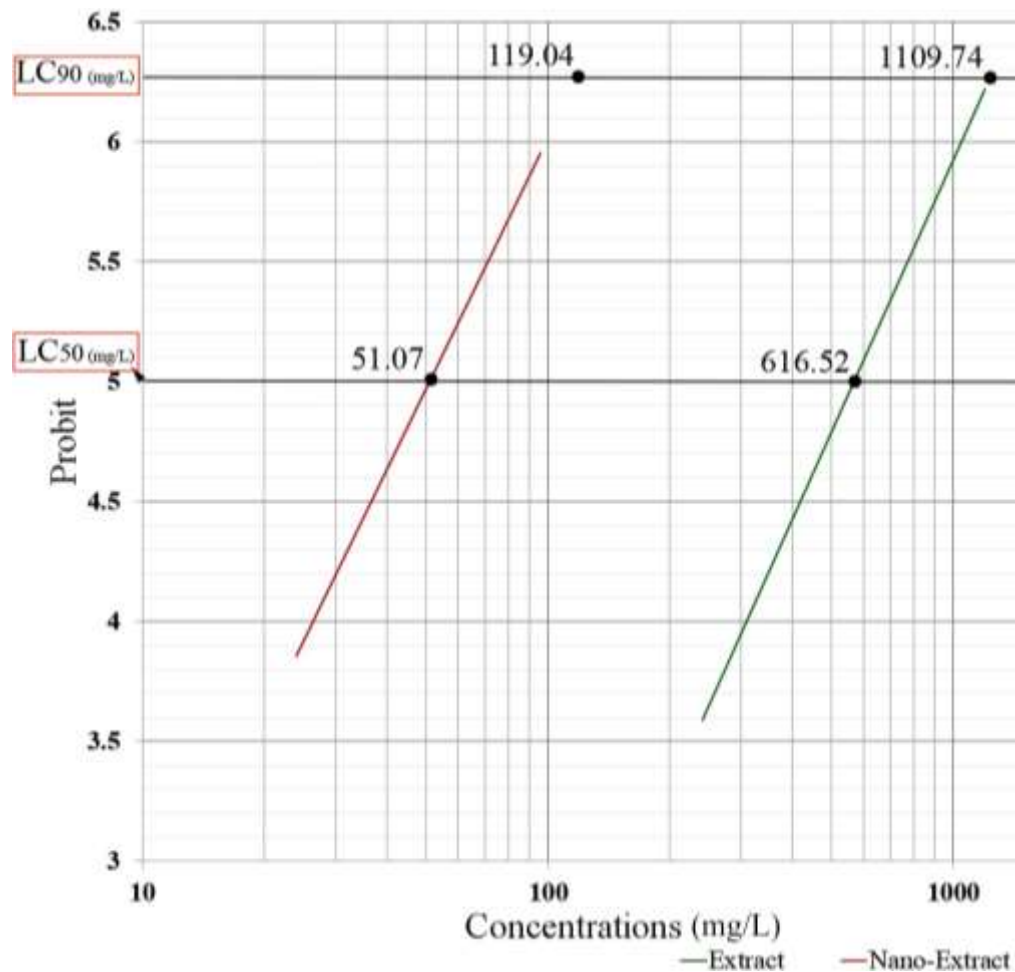


**Fig. 3.** Scanning electron microscope (SEM) image of silver nanoparticles synthesized using *Zataria multiflora* extract. Sample: Dried suspension of AgNPs coated on SEM stub. Particle shape: Mostly spherical, average diameter 24–63 nm. Device: FESEM / ZEISS / Sigma VP

**Table 2.** The repellency effect of *Zataria multiflora* extract and AgNPs against *Anopheles stephensi*, Shiraz City, Fars Province, Iran

| Compound | B±SE      | ED <sub>50</sub> (µg/cm <sup>2</sup> )±95 % CL | ED <sub>90</sub> (µg/cm <sup>2</sup> )±95 % CL | Df |
|----------|-----------|------------------------------------------------|------------------------------------------------|----|
| extract  | 2.84±0.32 | 0.00031±0.00005                                | 0.0018±0.0002                                  | 2  |
| AgNPs    | 1.76±0.21 | 0.000072±0.00001                               | 0.00048±0.00006                                | 2  |

**B:** Regression Coefficient, **SE:** Standard Error, **ED<sub>50</sub>:** Effective Dose to repel 50% of mosquitoes, **ED<sub>90</sub>:** Effective Dose to repel 90% of mosquitoes, **CL:** Confidence Limits, **DF:** Degrees of Freedom

**Fig. 4.** Probit regression lines of *Anopheles stephensi* larvae exposed to *Zataria multiflora* extract and silver nanoparticles, Shiraz City, Fars Province, Iran

## Discussion

### Larvicidal activity of plant extracts and AgNPs

Our study showed that the plant crude extract had weak larvicidal activity (LC<sub>50</sub> at 616.52 µg/mL and LC<sub>90</sub> at 1109.74 µg/mL). In contrast, the synthesized AgNPs showed signif-

icantly greater efficacy (>10 folds lower values), with LC<sub>50</sub> and LC<sub>90</sub> values of 51.07 µg/mL and 119.04 µg/mL, respectively, demonstrating stronger activity against mosquito larvae.

Marimuthu et al. (43) found the synthetic AgNPs of *Mimosa pudica* had the highest mortality rate among the synthetic silver nanoparticles against *Anopheles subpictus* larvae. Govindarajan et al. (44) investigated the effect of the methanolic extract of *Cassia fistul* against *An. stephensi* and showed that the LC<sub>50</sub> value of this substance is 17.97 µg/mL. A study on *Nucifera* leaves reported that the maximum potency larvicidal of *An. subpictus* was observed with synthetic AgNPs. The LC<sub>50</sub> value was 0.69 ppm, while the LC<sub>90</sub> value was 2.15 (45). Salunkhe found that the synthetic AgNPs with the fungus *Cochliobolus lunatus* were effective against *An. stephensi* (31). Another study showed that *Acacia cassia* using synthesized silver nanoparticles has a high larvicidal effect on *Ae. albopictus* (24). Muthukumar et al. (30) synthesized AgNPs using crude aqueous extracts of *Cassia roxburghii* and tested their efficacy against *An. stephensi*. The LC<sub>50</sub> value of 26.35 ppm, was calculated. The results of Patil et al. (31)'s study with AgNPs synthesized using *Plumeria Rubra* showed that the LC<sub>50</sub> value was 1.82 ppm against *Aedes aegypti*. Elumalai et al. (33) synthesized AgNPs using *Leucas aspera* and tested their efficacy against Culicidae mosquitoes. The most lethal effect was observed on *Ae. aegypti* and the least lethal effect was observed on *Cx. quinquefasciatus*. A study was conducted using crude aqueous extracts and synthesized AgNPs of *Gmelina asiatica* on *Cx. quinquefasciatus* and the results indicated that the lethal effect of AgNPs is greater than the crude extract (21). In general, the results of the conducted studies confirm the results of the present study. AgNPs were synthesized from *Z. multiflora* against the malaria vector *An. stephensi* for the first time in Iran and showed comparable activity against malaria vectors compared to crude extracts. However, the lethal dose was higher compared to other plants.

### Repellent activity of plant extracts and AgNPs

The study also evaluated the repellent activity of *Z. multiflora* extract and its AgNP

formulation. The extract showed an ED<sub>50</sub> of 0.00031 µg/cm<sup>2</sup> and ED<sub>90</sub> of 0.0018 µg/cm<sup>2</sup>. These results suggest considerable repellent capacity, though comparisons with commercial repellents such as DEET remain necessary.

The use of repellents represents a practical and cost-effective approach to mitigate the transmission of arthropod-borne diseases to humans. Among the various formulations available, those containing DEET (N, N-diethyl-3-methylbenzamide) are predominant due to their proven efficacy against mosquitoes and other biting insects (46). Despite its effectiveness, DEET application can result in a spectrum of human toxicity reactions, ranging from mild to severe (47). Consequently, there has been substantial research interest in developing plant-derived repellents as safer alternatives to DEET (48–49).

For instance, a study by Kelidari et al. (50) evaluated the repellent effect of a *Z. multiflora* nano formulation, which provided significantly extended protection (93±5 minutes) compared to an unformulated essential oil (29±2 minutes). This finding underscores the potential of botanical-based formulations as effective repellents. *Citronella* oil extracted from *Cymbopogon nardus* has also been extensively researched for its mosquito-repellent properties, demonstrating efficacy across various concentrations (49, 51). It is used either alone or in combination with other natural or commercial products to enhance efficacy against mosquitoes (52).

Beyond repellents, biological control measures for mosquitoes include plant extracts and essential oils, which serve as repellents, ovicides, larvicides, and adulticides. However, their effectiveness varies depending on the species and parts of the plant used (53). Our study demonstrated repellent activity against the malaria vector *An. stephensi* using extracts and silver nanoparticles synthesized from *Z. multiflora*.

In summary, while DEET remains a standard in mosquito repellents despite associated risks, ongoing research into botanical alterna-

tives highlights promising avenues for safer and effective repellent strategies against arthropod-borne diseases.

### Characterization of synthesized AgNPs

The SEM analysis showed that the synthesized nanoparticles were small and spherical, consistent with prior findings (54), where plant-mediated AgNPs typically assume this morphology. Particle shape and size significantly influence bioactivity, with smaller particles often offering greater surface area and interaction with target organisms. XRD analysis confirmed the crystalline nature of the AgNPs, revealing prominent peaks at  $2\theta$  28.35°, 40.56°, and 73.88°, indicating a globular structure (55). The sharp diffraction lines reflect successful nanoparticle formation and the presence of a bioorganic phase, which can enhance stability and biocompatibility.

### Environmental and practical implications

This study reinforces the potential of eco-friendly nanoparticle synthesis for vector control. Unlike conventional chemical pesticides, plant-mediated synthesis is cost-effective, requires less energy, and minimizes environmental pollution. However, deploying AgNPs at scale requires assessment of environmental risks, including toxicity to non-target organisms and accumulation in aquatic systems. Future research should address these risks and investigate biodegradable or hybrid formulations that retain efficacy while ensuring environmental safety.

### Study limitations

Some limitations of the study include: A lack of field validation for repellent or larvicidal efficacy under natural conditions. Limited comparison with standard chemical agents beyond DEET. The need for more detailed formulation and stability testing of AgNP-based repellents.

### Implications and future directions

The findings support the integration of bo-

tanical-based nanoparticle formulations into vector control strategies. Future studies should explore: Broader-spectrum activity against multiple mosquito species. The identification of active compounds responsible for larvicidal and repellent effects. Scalable production models suitable for public health implementation.

## Conclusion

This study demonstrated that silver nanoparticles synthesized using *Z. multiflora* possess significant larvicidal and repellent activities against *An. stephensi*. While not as potent as some AgNPs derived from other botanicals, their eco-friendly synthesis and biological efficacy highlight their potential as alternatives to conventional pesticides and synthetic repellents. These findings contribute to the growing interest in plant-based nanotechnology for sustainable vector control and warrant further exploration in real-world settings.

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## Ethical considerations

This study was approved by the research ethics committee of the School of Health, Shiraz University of Medical Sciences, and the ethical code number is IR.SUMS.REC.1398.978.

## Conflicts of interest statement

The authors declare that there is no conflict of interest.

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